### S.P.Mandali's Ramnarain Ruia Autonomous College



Syllabus for F.Y. B.Sc

Program: B.Sc.

Course: Biochemistry (RUSBCH)

(Credit Based Semester and Grading System with effect from the academic year 2019–2020)

#### Semester I

<b>Course Code</b>	Unit	Topics	Credits	Lectures	
Paper I - Biomolecules and Nutrition					
	I	Water and Biochemistry of minerals		15	
RUSBCH101	II	Nucleic acids	2	15	
	III	Carbohydrates		15	
	Paper II - Cell biology, Physiology and Microbiology				
	I	The cell		15	
RUSBCH102	II	Nucleus and cell cycle	2	15	
	III	Microscopy		15	
RUSBCHP101 Practi		cals based on course in theory -			
KUSBCHF 101	RUSB	3CH101	2		
RUSBCHP102	Praction	cals based on course in theory -	<b>5</b>		
KUSDCHP102	RUSB	3CH102			

#### Semester

<b>Course Code</b>	Unit	Topics		Credits	Lectures
Paper I - Biomolecules and Nutrition					
	I	Lipids			15
RUSBCH201	II	Amino acids and proteins		2	15
	III	Nutrition			15
Paper II – Introduction to cell biology, Physiology and Microbiology					
	I	Physiology of digestion and absorption	on		15
RUSBCH202	II	Physiology of respiration and excreti	on	2	15
	Ш	Microbiology			15
RUSBCHP201	Praction	als based on course in theory –			
KUSBCHF 201	RUSB	CH201		2	
RUSBCHP202	Praction	als based on course in theory –		2	
KUSBCHF202	RUSB	CH202			

Course Code: RUSBCH

Course Title:Biochemistry

Academic year 2019-20

#### **Learning Objectives:**

The overall goal of this FYBSc course is for the student to gain a basic working knowledge of biochemical concepts and techniques which will be necessary for future scientific endeavors.

#### **Learning Outcomes:**

Upon completion of the FYBSc course, the students are able to understand the following:

- 1) The basic scientific terms in the field of Biochemistry.
- 2) Enumeration of the biochemical functions of water, acids, bases & buffers.
- 3) The chemistry & structures of biomolecules (Carbohydrates, Proteins and Lipids), their classification and functions in living organism, structure-function relationship of biomolecules with their importance at molecular level. As these basic concepts form the basis for understanding metabolic fate of different biomolecules at the TYBSc level.
- 4) The Concept of the origin of life, basic cell structure and functions of cell organelles which is important for cytogenetics study and techniques associated with it.
- 5) An introduction to Microscopy which is included in the first semester of FYBSc. They will gain expertise to handle the microscope, helping them with the various microbial staining techniques which are a part of their practicals as well. This will also help them to understand the basics of microbiology included in the second semester.
- 6) The nucleic acids topic will lay the foundation to introduce them to the field of genetics.
- 7) Knowledge of physiological processes (Digestion, Absorption, Excretion etc.) and nutrition will enable them to understand metabolic and nutritional needs of the body which forms the basis of clinical and nutritional biochemistry.
- 8) All the practicals have been rearranged in accordance with the theory papers at each semester.
- 9) Students will learn to examine, assess, interpret and communicate data acquired performing laboratory experiments related to biochemistry.

#### **Detailed Syllabus**

		SEMESTER I	
Course C	ode	Title	Credits
RUSBCH1	101	Biomolecules and Nutrition	02
Unit I	Wat	er and Biochemistry of minerals	15 lectures
1.1 1.1.1 1.1.2 1.2	Hydr Effec	er, its structure and biological significance, rogen bonding and its importance, Water as a universal solvent. et of water on Biomolecules	100
1.2.1 1.2.2 1.3 1.4 1.5 1.5.1	Polar Weal Ioniz Bioc	opy and dissolution of solute r and non-polar compounds k interactions of biomolecules in aqueous solutions ration of water, weak acids and weak bases hemistry of minerals eral functions, classification	
1.5.2 Unit II	Mag	nanism of absorption and functions of – Calcium, Phosphorus, nesium, Sodium, Potassium, Iron, Iodine, Selenium eic Acids	15 lectures
2.1		eture - Purine & Pyrimidine bases, ribose, deoxyribose, nucleosides and	
2.2	nucle with RNA	ectides (ATP, CTP, GTP, TTP, UTP) Formation of polynucleotide strand its shorthand Representation as (various types in prokaryotes and eukaryotes) mRNA, hnRNA, rRNA, NA & snoRNA - general account, tRNA - clover leaf model, Ribozymes	
2.3 2.3.1	DNA Phys		
2.3.2	Phys (Viso dena	ical properties of DNA - Effect of heat on physical properties of DNA cosity, buoyant density, UV absorption), Hypochromism, hyperchromism, turation of DNA.	
2.3.3 2.4	Cent	tions of nucleic acids (with DPA and Orcinol) ral Dogma of Life (Definitions: Replication, Transcription, Translation & Transcription)	
Unit III	Carl	oohydrates	15 lectures
3.1 3.2 3.2.1 3.2.2 3.2.3	polys Mon- Class Struc ribos		
	a) Pl	erties: hysical- isomerism D & L, optical; optical; epimers: anomers hemical reactions –	

a) Composition: homo & hetero. with examples b) Storage: starch and glycogen - action of amylase on starch c) Structural: cellulose. Chitin  PRACTICALS  RUSBCHP101 PRACTICAL I Credit  1) Preparation & Standardization of laboratory reagents
PRACTICALS  RUSBCHP101 PRACTICAL I Credit
RUSBCHP101 PRACTICAL I Credit
RUSBCHP101 PRACTICAL I Credit
3) Qualitative tests for carbohydrates –  a) Monosaccharides - glucose & fructose b) Disaccharides - lactose, maltose, sucrose c) Polysaccharides - starch, dextrin 4) Qualitative tests to detect the unknown carbohydrates from the given solution 5) Qualitative test for nucleic acids

		SEMESTER I	
Course C	ode	Title	Credits
RUSBCH	102	Cell biology, Physiology and Microbiology	02
Unit I	The	Cell	15 lectures
1.1	Struc	etural organization of cell	C
1.1.1		aryotic, eukaryotic and yeast cells – A comparative overview	
		wall structure (plant), cell membrane (fluid mosaic model)	
1.2	Cyto	skeleton: microtubules & microfilaments	
1.3	Cell	organelles:	
1.3.1		chondrion: Organization & function of the mitochondria, mitochondrial	
1 2 2	geno		7
1.3.2		roplast: Structure & function of the chloroplast, the chloroplast genome,	
1.3.3		plastids some, ER, Golgi apparatus: Structure & Function of Ribosome, ER, Golgi	
1.5.5		ratus,	
1.3.4		xisome & Lysosome: Peroxisome function & assembly (in brief) and	
		some structure & function	
	•		
Unit II	Nucl	eus and cell cycle	15 lectures
2.1	Nucl	eus: Structure & Function of the nucleus,	
2.1.1		ear envelope – Nuclear membranes, perinuclear space, nuclear pores,	
		lus, central granule, fibrous laminnae. Functions of nuclear envelop	
2.1.2		eolus – Structure, types and its functions	
2.3		sis - Phases of mitosis, Cytokinesis	
2.3.1		ificance of mitosis	
2.3.2	Brea	kdown of nuclear envelope	
2.3.3	The s	spindle, Kinetochore	
2.3.4	Anap	phasic movements	
2.4	Meio	osis ( Types – zygotic, gametic and sporic)	
2.4.1		es of meiosis	
2.5	Cell	cycle and its regulation (cyclins & CDKs)	
Unit III	Micr	roscopy	15 lectures
3.1	Intro	duction and basic concept of Magnification, Resolving power, Numerical	
		ure, Limit of resolution, refractive index and role and RI of oil	
3.2	Parts	and functions of Compound microscope	
3.3	Light	t microscope- Bright Field, Dark field, Phase contrast, Fluorescence	
	micro	oscopy( Immunofluorescence and FISH), Confocal Microscopy	
3.4		tron Microscopy- SEM,TEM	
3.5	Elect	tron tomography	
3.6	Integ	grated Microscopy (only principle)	
		PRACTICALS	
RUSBCHI	P102	PRACTICAL II	Credits 01
		Parts and maintenance of a Microscope	
		1, 1 and and maintenance of a microscope	

- 2) Gram Staining
- 3) A study of electron micrographs of cell organelles
- 4) Permanent Slides of Bacilli, Spirilla, cocci, Rhizopus
- 5) Nucleic acid Staining
- 6) Effect of hypo, hyper, isotonic solutions on cells of onion peel
- 7) Permanent slides of Mitosis and Meiosis
- 8) Qualitative Analysis of Minerals -Sodium, Potassium, Phosphorus, Iron, Chloride, Nitrate



		SEMESTER II	
Course (	Code	Title	Credits
RUSBCH	<b>I201</b>	Biomolecules and Nutrition	02
Unit I	Lipi	ds	15 lectures
1.1	Defin	nition, classification (Bloor's) and functions of Lipids	C
1.2		acids and Triacylglvcerol	
1.2.1		sification &Chemistry,	040
		rated fatty acids - classification of C2 to C20: even carbon: Common and	16
		AC names. Unsaturated fatty acids MUFA, PUFA (2.3.4 double bonds) ga - 3.6.9 fatty acids. Triacyl glycerol - simple and mixed - names and	
	struc		
1.2.2		nical Reactions of fats	
1.2.2		nification, Iodination, Ozonolysis, Auto-oxidation,	
	_	on of heat on glycerol and choline,	
		idity Definition & significance - Acid number, Saponification number,	
		e number, Reichert - Meissel number	
1.3	Com	pound Lipids	
		tions of glycerophospholipids (PE.PC.PL)	
		phosphingolipids (ceramide, sphingomyelin), Glycolipids /Cerebrosides	
		o&galactocerebrosides)	
1.4	Stero	oids Cholesterol structure and biochemical significance	
		<u> </u>	4
Unit II	Ami	no acids and proteins	15 lectures
2.1		no acids	
2.1.1		no acid structure- D & L forms of all 20 amino acids	
2.1.2		iled classification based on polarity: essential & non-essential amino acids,	
2.1.2		ter ion, pI of amino acids, amino acids as ampholytes	
2.1.3		ical and chemical properties- Chemical reactions of amino acids with	
		ydrin, Sanger's reagent, Edman's reagent and Dansyl chloride. Functions of	
2.2		o acids (in brief). des and Proteins	
2.2.1		C- APS classification on the basis of shape, function and physical properties	
2.2.1		lubilities (Simple, conjugated & derived proteins). Nutritional classification	
		nplete & incomplete proteins)	
2.2.2	,	erties of proteins. Primary structure -Formation and characterization of the	
		de bond	
2.2.3		ndary structure -Alpha helix and beta sheet	
2.2.4		ary and Quaternary structures- an introduction with one example of each	
2.2.5	Prote	in denaturation	
	_		
Unit III	Nutr	rition	15 lectures
3.1	Nutri	tion & Energy Supply - Calorie, Joule, Respiratory Quotient	
3.2		calorimetry - calorific value determination by Bomb calorimeter, calorific	
		es of proximate principles,	
3.3		zation of energy in man - BMR (Definition, factors affecting BMR,	
		ficance of BMR in clinical diagnosis), SDA/DIT -General concept and	
	signi	ficance, Energy requirement of individuals for various activities-	

3.5	Asses	tional importance of Carbohydrates, Proteins, Lipids and Fiber ssment of nutritive value of proteins - Chemical score of amino acids,	
		in Deficiency Corrected Amino Acid Score, BV, PER, NPU	
3.6		ept of BMI, RDA	
3.7		aced diet	
3.8		tional disorders	
3.9	Nume	erical problems based on above concepts	
		PRACTICALS	6
RUSBCI	HP201	PRACTICAL I	Cred
		1) Quantitative estimation of normality of FAS	
		2) Qualitative tests for amino acids	
		3) Qualitative tests for proteins	
		4) Tests to detect unknown proteins	
		5) Qualitative tests for lipids	
		6) Estimation of Saponification value	
		7) Estimation of Acid value	
		8) Estimation of vitamin C by dye method	
		9) Demonstration of Analytical balance	

		SEMESTER II	
Course C	ode	Title	Credits
RUSBCH	202	Introduction to Cell biology, Physiology and Microbiology	02
Unit I	Phys	iology of digestion and absorption	15 lectures
1.1 1.2	Orga	and Functions of gastro intestinal tract (GIT) ns and Glands associated with GIT Secretions and Juices of GIT (Saliva, ric juice, Intestinal juice, pancreatic and Bile juice)	60
1.3 1.4	_	stion and Mechanism of Absorption of carbohydrates stion and Mechanism of Absorption of Lipids	
1.5	Dige	stion andMechanism Absorption of Proteins	
1.6	Disor	rders - Peptic ulcer, Lactose Intolerance, Celiac disease, Pancreatitis	
Unit II	Phys	iology of respiration and excretion	15 lectures
2.1 2.2 2.3	Breat	iratory system, thing - inspiration and expiration, position of air and partial pressure of gases	
2.4 2.4.1	•	ical exchange of gases sport of oxygen	
2.4.2		sport of carbon dioxide	
2.5	occuj	iratory disorders – Asthma, pharyngitis, laryngitis, hay fever, pneumonia, pational lung disease (silicosis & asbestosis), cyanosis, respiratory acidosis llkalosis	
2.6	Excre		
2.6.1	funct	ture of the nephron: Bowman's capsule & glomerulus - Structure & ion, (ultrafiltration, pressures involved, GFR, regulation of GFR); Renal e - structure & function (proximal and distal convoluted tubules and Henle's	
2.6.2		e formation: Reabsorption / Secretion of glucose, Na+, K+. HCO3 nd H+: renal threshold,	
2.6.3	Norn	nal & Abnormal constituents of urine, Excretory disorder: Nephritis	
Unit III		robiology	15 lectures
3.1		rical background (contributions or Leeuwenhoek. Pasteur ,etc ) and General	
3.2	Microbased	cteristics (size .shape. and structure) of Bacteria obial Taxonomy: Microbial species and strains. Classification of bacteria don morphology (shape and flagella). staining reaction, nutrition and	
13		me environment (extremophiles: Thermophiles, Psychrophiles, Halophiles, netotactic, Radiation resistant organisms: examples with their application)	
3.3		erial cell wall: Structure and function, components of peptidoglycan	
	frame	ework (structure of NAG & NAMA not necessary)	
3.4		ing methods (principles of staining & types or stains)and microscopic	
3.5	Micr	ification of bacteria obial Growth - Growth Curve, Mathematical expression, Synchronous th, Generation time	
3.6	_	are media (N, C, Special requirements), Natural and Synthetic media	

RUSBCHP202  1) Study of Human Digestive, Respiratory & Excretory System with the help of diagrams  2) Estimation of total acidity of gastric juice  3) Urine analysis: Normal Constituents: Urea, Uric acid, Chloride Abnormal Constituents: Glucose, Protein  4) Titrable acidity using Neutral red or Phenol red  5) Qualitative tests for the detection of functional groups  6) Capsule Staining  7) Endospore Staining  8) Lipid Staining	1) Study of Human Digestive, Respiratory & Excretory System with the help of diagrams 2) Estimation of total acidity of gastric juice 3) Urine analysis: Normal Constituents: Urea, Uric acid, Chloride Abnormal Constituents: Glucose, Protein 4) Titrable acidity using Neutral red or Phenol red 5) Qualitative tests for the detection of functional groups 6) Capsule Staining 7) Endospore Staining 8) Lipid Staining		PRACTICALS	
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8) Lipid Staining	8) Lipid Staining			9
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Solution				

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#### MODALITY OF ASSESSMENT

#### **Theory Examination Pattern:**

#### A) Internal Assessment - 40% :40 marks.

Sr No	Evaluation type	Marks
1	One Assignment/poster presentation/Quiz	20
2	One class Test (multiple choice questions / subjective)	20

#### B) External examination - 60 %

#### **Semester End Theory Assessment - 60 marks**

- i. Duration These examinations shall be of **2 hours** duration.
- ii. Paper Pattern:
  - 1. There shall be **04** questions each of **15** marks. On each unit there will be one question &last question will be based on all 3 units.
  - 2. All questions shall be compulsory with internal choice within the questions.

Questions	Options	Marks	Questions on
Q.1)A)	Any 2 out of 3	04	Unit I
Q.1)B)	Any 2 out of 3 OR Any 4 out of 6	06	
Q.1)C)	Any 1 out of 2	05	
Q.2)A)	Any 2 out of 3	04	Unit II
Q.2)B)	Any 2 out of 3 OR Any 4 out of 6	06	
Q.2)C)	Any 1 out of 2	05	
Q.3)A)	Any 2 out of 3	04	Unit III
Q.3)B)	Any 2 out of 3 OR Any 4 out of 6	06	
Q.3)C)	Any 1 out of 2	05	
Q.4)A)	Any 1 out of 2	04	Unit I, II, III
Q.4)B)	Any 1 out of 2	04	
Q.4)C)	Any 1 out of 2	04	

Q.4)D)	Any 3 out of 5	03	
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#### **Practical Examination Pattern:**

#### (A)Internal Examination:

Heading	Practical I & II
Journal	05
Test	15
Total	20

#### (B) External (Semester end practical examination):

Particulars	Practical I & II
Laboratory work	25
Viva	5
Total	30

#### PRACTICAL BOOK/JOURNAL

The students are required to present a duly certified journal for appearing at the practical examination, failing which they will not be allowed to appear for the examination.

In case of loss of Journal and/ or Report, a Lost Certificate should be obtained from Head/ Coordinator / Incharge of the department; failing which the student will not be allowed to appear for the practical examination.

### Overall Examination and Marks Distribution Pattern Semester – I & II

Course	RUSE	BCH101		RUSE	SCH102		Grand Total
	Internal	External	Total	Internal	External	Total	
Theory	40	60	100	40	60	100	200
Practicals	20	30	50	20	30	50	100

Course	RUSBCH201		RUSBCH202			Grand	
							Total
	Internal	External	Total	Internal	External	Total	
Theory	40	60	100	40	60	100	200
Practicals	20	30	50	20	30	50	100

Resolution No. AC/II(18-19).2.RUS2

# S.P.Mandali's Ramnarain Ruia Autonomous College



Syllabus for S.Y.B.Sc.

Program: B.Sc.

Course: Biochemistry (RUSBCH)

# (Credit Based Semester and Grading System with effect from the academic year 2019–2020)

#### **Semester III**

<b>Course Code</b>	Unit	Topics	Credits	Lectures
Pa	per I –	Enzymology, Physiology & Tools of Bioche	emistry	
	I	Enzymes and coenzymes	( )	15
DIJCDCII201	II	Plant Biochemistry	2	15
RUSBCH301	III	Acids, Bases, Buffers and Ionic Equilibria	<b>5</b> 2	15
		& pH meter		13
	Paper	· II - Fundamentals of Genetics and Physiol	logy	
	I	Genetics: I		15
RUSBCH302	II	Genetics: II	2	15
	III	Transport mechanisms and haematopoesis		15
	Paper	III -Industrial Biotechnology & Pharmaco	logy	
	I	Sterilization & Disinfection Techniques		15
RUSBCH303		and Cell culture		13
RUSBCH303	II	Fermentation and Downstream processing		15
	III	Industrial Biotechnology		15
RUSBCHP301	Practi	cals based on the course in theory –		
KUSBCHF301	RUSE	3CH301		
RUSBCHP302		cals based on the course in theory –	3	
ROSDCIII 302	RUSE	3CH302	3	
RUSBCHP303		cals based on the course in theory –		
ROSDCIII 303	RUSE	CH303		
RUSBCH303				
10)				

#### **Semester IV**

<b>Course Code</b>	Unit	Topics	Credits	Lecture
Pa	per I - Enz	ymology, Physiology & Tools of Bioch	emistry	
	I	Physicochemical Principles	U	15
RUSBCH401	II	Hormones	2	15
	III	Tools of Biochemistry	3	15
	Paper II -	<b>Fundamentals of Genetics and Physic</b>	logy	
	I	Locomotion & Movement		15
RUSBCH402	II	Neurophysiology	2	15
	III	Body Fluids		15
	Paper III -	Industrial Biotechnology & Pharmac	ology	
	I	Trends in Biotechnology		15
DUCDCUA02	II	Introduction to Pharmacology		15
RUSBCH403	III	Pharmacodynamics and Adverse	- 2	1.5
		Drug Reactions (ADR)		15
RUSBCHP401	Practicals 1	based on the course in theory –		
KUSBCHF401	RUSBCH4	401		
RUSBCHP402		based on the course in theory –	3	
ROSDCIII 402	RUSBCH	402		
RUSBCHP403		based on the course in theory –		
ROSDCIII 403	RUSBCH <sup>2</sup>	403 and Project work		

**Course Code:** RUSBCH

Course Title: Biochemistry

Academic year 2019-20

#### **Learning Objectives:**

The overall goal of this SYBSc course is to introduce the student to the fields of enzymology, plant biochemistry, genetics, endocrinology, clinical biochemistry, industrial biotechnology, and pharmacology.

#### **Learning Outcomes:**

Upon completion of the SYBSc course, the students would understand the following:

- 1) Enzymology which forms the core of Biochemistry. Enzyme immobilization to study a different aspect of enzyme kinetics.
- 2) Plant Biochemistry which will introduce the students to metabolism and enable them to study metabolism in plants.
- 3) Acids, Bases, Buffers and Ionic Equilibria& pH meter; hormones; membrane biochemistry and physicochemical principles which will help them to further understand the biochemical processes of the body.
- 4) Mendelian Genetics, variations over Mendelian Genetics, Genes and Chromosomes topics which will introduce them to the field of genetics.
- 5) Haematopoiesis and Body fluids topics to help them understand the basics of clinical biochemistry.
- 6) Concept of sterilization and disinfection, ATC, PTC & fermentation to enable them to understand the various microbial processes & techniques.
- 7) Tools of Biochemistry which includes basic instrumentation like Colorimetry, Spectrophotometry, Flame photometry, Fluorimetry and Flow Cytometry.

- 8) The applied aspects of Biochemistry through Biotechnology, Industrial Biosynthesis, Bioresources and their management.
- 9) General pharmacology, Pharmacodynamics and Adverse Drug Reactions (ADRs) will introduce the students to pharmacology.
- 10) All the practicals have been rearranged in accordance with the theory papers at each semester.
- 11) New experiments are added to cover the practical aspects of the newly added theoretical topics. E.g. colorimetric experiments like Glucose by DNSA method, Proteins by Biuret method and Demonstration of spectrophotometer.
- 12) Research project has been introduced in Semester IV Practical III to inculcate research culture in the students. This will familiarize them with Research methodology i.e. atal c alation of alat reference work, experimental work, analysis of experimental data, interpretation of results obtained, writing of project or work and compilation of bibliography in proper

#### **Detailed Syllabus**

			SEMESTER III					
	Cours	se Code	Title	Credits				
	RUSBCH301 Enzymology, Physiology & Tools of Biochemistry							
Un		zymes and C						
1.1			Enzyme, Apoenzyme, Holoenzyme, Prosthetic group, Active site, Enzyme	15 Lectures				
		specificity, Turnover number, Specific activity, Katal, IU, Coenzyme and Cofactor						
1.2			ssification upto one digit.					
1.3			ficity: Fischer's, lock & key and Koshland's, induced fit theories					
1.4			ergy, Mechanism of Enzyme action (concept of active site, single and bi-					
			tion), Factors affecting enzyme activity – substrate concentration, pH,					
		emperature						
1.5			etics – Derivation of Michaelis - Menten equation and Lineweaver Burke plot					
			trate reactions and numerical problems based on them					
1.6		•	bition – Reversible and Irreversible; Competitive and Non competitive, one					
		xample of ea						
1.7	7 P	Problems on e	entire Unit					
		lant Biochen		T				
2.1			Substances- Structure and Function of- Auxins, Gibberllins, Cytokininis,	15 Lectures				
			Abscissic Acid					
2.2		hotosynthesi						
			reactions, Z scheme and electron carriers, photophosphorylation [linear and					
			respiration, Photoperiodism Calvin cycle – schematic with enzymes, C <sub>4</sub> and					
		CAM pathway						
2.3			etabolites of plants – Nitrogen containing compounds (Alkaloids), Terpenes					
			ompounds – An introduction to Shikimic acid pathway, Mevalonic acid					
	p	athway, MEF	P Pathway					
TT-4	2711	· d. D	D. 66					
3.1			Buffers and Ionic Equilibria & pH meter H, pK, pKw, Isoelectric pH, Buffer, Buffering Capacity	15 Lectures				
3.2			Ionic product of water, Hendersen–Hasselbalch equation, Relation between	13 Lectures				
3.2								
3.3	_	pI, pKa1 and pKa2 for a neutral, acidic and basic amino acid Titration and Ionization of Glycine, Lysine and Aspartic acid; pKa and pI values of these						
] 3.3		amino acids.						
3.4			glass electrode					
3.5			Buffers: (Hb-HHb, Carbonate-Bicarbonate, Phosphate, and Protein)					
3.6		•	a above concepts					

		1
	PRACTICALS	
RUSBCHP301		Credits 01
	1) Extraction of β-Amylase from sweet potato	
	2) Extraction of Urease from horse gram	
	3) Extraction of Invertase from baker's yeast	
	4) Preparation of Immobilised Yeast and its use in determination of Invertase activity	34
	5) Determination of Achromic point of Amylase	
	6) Preparation of Buffers and measurement of pH using paper and pH meter	1,5
	7) Acid –Base titration of a Polyprotic acid	
	8) Determination of refractive index using portable refractometer.	

		SEMESTER III				
Cot	ırse Code	Title	Credits			
RU	SBCH302	Fundamentals of Genetics and Physiology	02			
Unit I	Genetics : I					
1.1		netics: Mendels experiments, Laws of inheritance	15 Lectures			
1.2		er Mendelian Genetics - Incomplete Dominance Co-Dominance,				
		les, Pleiotropy, Polygenics, Epistasis, Linked Genes, Sex-linked Genes,				
		al influences on Gene Expression (Hormones, Sex-limited & Sex-				
	, ,	Maternal Gene Effects				
1.3		n above concepts				
1.4		l abnormalities (Down's Syndrome, Klinefelter's Sydrome, Turner's				
	Syndrome, Cr	ri-du-chat syndrome, Philadelphia Chromosome)				
	<b>Genetics: II</b>					
2.1	Prokaryotic G	Senome: Circularity; Single origin	15 Lectures			
2.2	Eukaryotic ch	romosomes: Organization of DNA into chromosomes (upto Solenoid				
	structure)					
2.3		iling, Topoismerase, Chromatin structure, Euchromatin,				
	Heterochromatin, structure of condensed chromatin, Nucleosomes, [Centromere,					
	kinetochrome, telomere], Acetylation & deacetylation of histones, Role of Telomerase					
2.4	Comparison of	of chromosomal structure in prokaryotes and eukaryotes				
2.5	Transformation	on: Definition and transformation in S.pneumoniae				
2.6	Transduction	: Definition; Explain general features with one example				
2.7	Conjugation:	Mechanism, F+, F- and Hfr strain				
	I Transport n	nechanisms and Haematopoesis				
3.1	Transport med	chanism across Cell Membrane	15 Lectures			
3.1.1		eins and Carrier proteins				
3.1.2	Active transpo	ort (primary – Na+ & K+ pump; secondary – Glucose) & Passive				
$\mathcal{D}^{\prime}$		ple and facilitated diffusion) with suitable examples; concept of symport				
3.1.3	and antiport. l	Endocytosis and Exocytosis – with one example each.				
3.2	Haematopoes					
3.2.1	RBCs, WBCs	, granulocytes & platelets – their development, morphology and				
	maturation (in	,				
3.2.2	Biosynthesis	of haemoglobin, its regulation and metabolism				
		PRACTICALS				

RUSBCHP302		Credits 01
	1) Laws of Mendel	
	2) A study of Human Karyotypes	
	3) Effect of organic solvents on cell rupture	
	4) Isoelectric precipitation of casein	
	5) Recrystallization of Benzoic acid	
	6) Estimation of Ascorbic acid Iodometrically	
	7) Detection of generation of carbon dioxide during fermentation by	
	yeast	00
	8) Sugar fermentation test	
	9) Assignment – G banding, C banding, Q banding, of chromosomes	

			SEMESTER III					
	Course Code		Title	Credits				
F	RUSBCH303 Industrial Biotechnology & Pharmacology							
Unit 1	I Sterilization	1 & D	Disinfection techniques and Cell culture					
1.1			Disinfection techniques	15 Lectures				
	Physical Agent of sterilization - Temperature- Pressure (Hot Air Oven, Autoclave),							
			Gamma) (examples with mechanism) Chemical agents of sterilization -					
			ns, Formaldehyde					
1.2			ulture: Introduction, Requirements, Culture techniques (Contamination and					
		); Cu	lture media: Stem cell culture; Advantages; Applications-Hybridomas,					
1.0	vaccines	4.						
1.3			ture: Introduction; Plant breeding; Techniques for maintenance; Genetic					
			s: Callus regeneration, mutant selection from culture; Protoplast fusion,					
	Transformat	10n: 1	Applications					
<b>T</b> T • ( )	<u> </u>							
			and Downstream processing	15 Lectures				
2.1	Basics: Primary and secondary screening, preparation of inoculum, and fermentation media							
2.2	Fermentors: Types (Batch, Continuous and fluidized bed) and its construction and accessory							
2.2	equipment; Operation of a fermentor; sterilization, inoculation, aeration, agitation;							
2.3			cessing: Introduction, Separation of particles (solidliquid), Cell disruption,					
	extraction methods, concentration, Purification and resolution of mixtures, drying							
I Init	 	l Dia	tashnalagu					
3.1	III Industria		sis - Penicillin, Vit B12, Cheese, Amylase / Protease; Ethanol; Acetic Acid	15 Lectures				
3.1			ymes: Introduction; Methods of immobilization (entrapment, adsorption,	13 Lectures				
3.2		_	microencapsulation, cross-linking); Stabilization of soluble enzymes:					
			trate stabilization, Enzyme stabilization by polymer, salts, and Chemical					
			plications and Problems					
3.3			ares of biosensors; Types: Electrochemical, Thermometric, Optical,					
3.5			nole cell, Immunobiosensor; Construction and development, Applications					
3.4	Single cell proteins: Introduction; Bacterial proteins; Yeast proteins; fungal proteins; algal p							
	roteins; Economic aspect; Applications							
	10001115, 200							
			PRACTICALS					
RUSI	BCHP303			Credits 01				
		) De	emonstration of the Working of an Autoclave and a Hot Air Oven					
			erility Testing of Air by plate exposure technique					
L	l –	/	7 O TAR MARKET PROPERTY AND	l				

2)	
3)	Study of microbial growth curve using optical density
4)	Cell count in a culture medium using optical density
5)	A study of culture inoculation methods – Pour plate, Spread plate & Streak
	Plate
6)	Determination of minimum inhibitory concentration of any one
	disinfectant
7)	Flow sheet diagrams of industrial synthesis of a Vitamin, an Antibiotic, a
	Food item, an enzyme, and alcohol
8)	Demonstration of alcohol generated during fermentation by yeast
	SEMESTER IV

		SEMESTER IV				
Cour	se Code	Title	Credits			
RUSBCH401		Enzymology, Physiology & Tools of Biochemistry	02			
Unit I l	Physicoche	mical Principles				
1.1		and Osmosis	15 Lectures			
1.1.1		n of : solute concentration (ways of expressing it- mole, molal, normal, percent),				
		ionic strength,				
1.1.2		Diffusion & definition of diffusion coefficient (factors affecting solute diffusion in solution)				
1.1.3		Osmosis- VantHoffs law of osmotic pressure (state law & write mathematical expression- no				
		, mechanism of osmosis, Role of osmosis in physiology. Eg. Renal dialysis.				
		n of solute between two immiscible solvents				
1.2		nd Viscosity				
1.2.1		tate in relation to surface forces, surface area, electrical charge, precipitation and				
	flocculatio					
1.2.2		nsion and its measurement, factors affecting surface tension. Eg. Role of bile in				
1 2 2	digestion					
1.2.3		- definition, measurement; Donnan membrane equilibrium, relation between				
101		onnan equilibrium and osmotic pressure Isorption and its characteristics				
1.2.4	Adsorption	and its characteristics				
TI-si4 TT	II					
2.1	Hormones	of Hormones, hormone receptor, endocrine & exocrine glands	15 Lecture			
2.1		tion of hormones on the basis of:	13 Lecture			
2.2						
		of target tissue- autocrine, paracrine, endocrine try - One example for each sub class.				
2.3		organization of the mammalian endocrine system				
2.4		synthesis, secretion & physiological role of thyroxine and insulin (Synthesis from				
2.4		lin), Diabetes mellitus, Hypothyroidism (cretinism and myxedema),				
		pidism (goiter – simple & toxic)				
2.5		cal role of glucocorticoids, oxytocin & vasopressin, FSH, LH, Estrogen,				
		ne (Reproductive cycle)				
2.6		ction of steroid hormones and epinephrine. (amplification cascade with G proteins,				
		enylatecyclase, kinases)				
	,					
Unit II	I Tools of 1	Biochemistry	1			
3.1		otometry and Colorimetry	15 Lecture			
3.1.1		pert law, derivation, limitations, application – estimation of sugar(DNSA) and				
		uret); concepts of Lambda max; determination of molar extinction coefficient				

1) Adsorption of oxalic acid on activated charcoal 2) Determination of viscosity by Ostwald's Viscometer 3) Determination of density of sugar syrup 4) Demonstration of spectrophotometer 5) Estimation of glucose by DNSA method 6) Estimation of proteins by the Biuret method 7) Estimation of anthocyanin content in vegetable 8) Determination of consistency of juices / sauces /squashes / syrup using portable consistometer.	3.2 Principles and applications of – 3.2.1 Flame photometry 3.2.2 Fluorimetry 3.2.3 Flow cytometry, FRAP, FRET, FLIM  PRACTICALS  RUSBCHP401  1) Adsorption of oxalic acid on activated charcoal 2) Determination of viscosity by Ostwald's Viscometer 3) Determination of density of sugar syrup 4) Demonstration of spectrophotometer 5) Estimation of glucose by DNSA method 6) Estimation of proteins by the Biuret method 7) Estimation of anthocyanin content in vegetable 8) Determination of consistency of juices / sauces /squashes / syrup using	3.2		tions of Beer-Lambert law in the estimation of sugar [DNSA] and protein[Biuret]	
3.2.1 Flame photometry 3.2.2 Fluorimetry 3.2.3 Flow cytometry, FRAP, FRET, FLIM  PRACTICALS  RUSBCHP401  1) Adsorption of oxalic acid on activated charcoal 2) Determination of viscosity by Ostwald's Viscometer 3) Determination of density of sugar syrup 4) Demonstration of spectrophotometer 5) Estimation of glucose by DNSA method 6) Estimation of proteins by the Biuret method 7) Estimation of consistency of juices / sauces /squashes / syrup using portable consistometer.	3.2.1 Flame photometry 3.2.2 Fluorimetry 3.2.3 Flow cytometry, FRAP, FRET, FLIM  PRACTICALS  RUSBCHP401  1) Adsorption of oxalic acid on activated charcoal 2) Determination of viscosity by Ostwald's Viscometer 3) Determination of density of sugar syrup 4) Demonstration of spectrophotometer 5) Estimation of glucose by DNSA method 6) Estimation of proteins by the Biuret method 7) Estimation of anthocyanin content in vegetable 8) Determination of consistency of juices / sauces /squashes / syrup using portable consistometer.		Numerio	cals based on the above concepts	
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Section 2.2.3   Flow cytometry, FRAP, FRET, FLIM   PRACTICALS	Section 2013   Flow cytometry, FRAP, FRET, FLIM   PRACTICALS		Flame p	hotometry	
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		SEMESTER IV	
	ırse Code	Title	Credits
	SBCH402	Fundamentals of Genetics and Physiology	02
	Locomotion &		
1.1	Muscle contra		15 Lectures
1.1.1		cles – Smooth, Skeletal, Cardiac	
1.1.2		anization of a muscle fibre, myofibril.	
1.0		ns – Actin, Myosin, Troponin, Tropomyosin	
1.2		nd Relaxation of Muscles – mechanisms (Sliding filament theory)	
1.2.1		f contractions – eg twitch, tetanus, wave summation, Isotonic, Isometric	
		Muscle contraction plant movements	
1.2.2		ements – Xerochasy, Hydrochasy	
1.2.2	•	ents – Protoplasmic streaming, paratonic movements	
		nents – Chemotaxis, Phototaxis, Thermotaxis	
		nents – Chemo / geo / hydro / photo / thigmo tropism	
		nents – Seismonasty, Nyctynasty, Photonasty, Chemonasty,	
	Thermonasty	ients seismonasty, typetymasty, i notomasty, enemonasty,	
	•	movements - Ciliary, Amoeboid, Cyclosis	
		,	
Unit II	Neurophysio	logy	
2.1		em – Classification: CNS, PNS; Components : Neurons (3 types) and	15 Lectures
	Neuroglia (6 t	ypes) – structure and function, Axonal transport	
2.2		orane Potential, ion channels [voltage and ligand gated], Action	
		olarization, polarization and refraction period), propagation of action	
	* - ·	statory & continuous conduction)	
2.3		anatomy of a synapse ; Transmission at synapses – Electrical &	
		apses, Excitatory & Inhibitory post synaptic potentials, Agonists &	
2.4		Removal of Neurotransmiters	
		tters- acetylcholine and Catecholamines, GABA, Glutamate &	
	Aspartate – st	ructure and function	
<b>*</b>			
	I Body Fluids		15.7
3.1		ments of the body – ICF and ECF	15 Lectures
3.2		osition, characteristics and function; role of plasma proteins, Starlings	
2 2		ood clotting and factors involved [ no pathway]	
3.3	ые: Compos	sition, characteristics and function; storage	

3.4	Lymph:	Lymph : Composition, Formation and Circulation		
		PRACTICALS		
RUS	BCHP402		Credits 01	
		1) Permanent slides of Muscle Tissue		
		2) Demonstration of Plant Movements		
		3) Determination of Total WBC count: Haemocytometry	- 0	
		4) Determination of Total RBC count: Haemocytometry		
	5) Bleeding time			
		6) Clotting time		
		7) Effect of Hypotonic, Isotonic and Hypertonic solutions on RBC's		
		8) Bile Analysis: Detection of Bilirubin & Bile salts		
		9) Determination of acidity of Vinegar		

	SEMESTER IV					
C	Course Code Title					
R	USBCH403	Industrial Biotechnology & Pharmacology	02			
	Unit I Trends in Biotechnology					
1.1	Bioaugmentat	ion, Genetically Engineered Microbes	15 Lectures			
1.2	Bioremediation	Bioremediation: Introduction; Factors affecting bioremediation; Types; Types of				
	reactions (Ae	reactions (Aerobic, anaerobic, sequential); Biodegradation of – hydrocarbons,				
	xenobiotics, h	eavy metals (Uranium); Paper Pulp Industry.				
1.3	Biopesticides,	Biofungicide				
1.4	Bioherbicides	and agricultural antibiotics				
1.5	Biofertilizers					
1.6	Integrated Pes	at Management				
		to Pharmacology				
2.1	Scope of phar		15 Lectures			
2.2		sification and Nomenclature of drugs				
2.3	_	and routes of drug administration; Factors affecting dosage and drug				
	delivery					
2.4		etics: LD 50, ED 50 Half Life, Loading dose, Maintenance dose				
		of terms only), Therapeutic dose, Therapeutic Index, Drug plasma				
	concentration	concentration, Volume of distribution, Clearance				
		ynamics and Adverse Drug Reactions (ADR)				
3.1	Pharmacodyn		15 Lectures			
3.1.1						
3.1.2		Drug Receptor Interaction – Receptor Theory of Drug Action, Location of Drug				
		Receptor-G-Coupled Protein receptors				
3.1.3						
3.1.4						
3.1.5						
3.1.6	•	Affinity and Intrinsic Activity				
3.1.7	•	Intensity of Drug Response – Potency and Efficacy				
3.1.8	Combined Effects of Drug – Synergism, Antagonism					
3.2		Adverse Drug Reactions (ADR)				
3.2.1		1 Types of ADR				
3.2.2		age, Drug dependence, Over dosage, Acute poisoning				
3.2.3	General Princ	iples of Management of Poisoning				

RUSBCHP403	PRACTICALS	Credits 01
Nesseal ide	<ol> <li>Determination of the amylase exoenzyme secreted by the microorganisms</li> <li>Determination of the zone of inhibition of microorganisms using the agar well method and disc diffusion method</li> <li>Study of antimicrobial activity of different plant sources.</li> <li>Qualitative determination of reducing sugars formed by starch hydrolyzing micro-organisms.</li> </ol>	600
	PROJECT WORK	
	GUIDELINE TO CARRY OUT PROJECTWORK  1. The main purpose of introduction of Project Work at SYBSc is to inculcate research culture at under-graduation level. It will also make the students familiar with Research Methodology i.e. reference work, experimental work, analysis of experimental data, interpretation of results obtained, writing of project work and compilation of bibliography in proper order.	
	2. Each student individually or in a group shall complete a small research project during their academic year of SYBSc. However, the initial reference work for the project can be started after the conclusion of FYBSc Semester II examination and summer vacation to SYBSc.	
	3. <b>Nature of Research Project:-</b> Experimental-based involving laboratory analytical work will be considered as the Research Project.	
	4. <b>Duration of Project work:-</b> Using the infrastructure available in the Biochemistry Department, Ramnarain Ruia Autonomous College, the duration to complete the project work will be from the commencement of the project work till the end of January of SYBSc (Sem IV) academic year.	
	5. Schedule for Submission of project Work:-Experimental work must be completed and the report on the same (2 Copies) will have to be submitted by the end of January of SYBSc (Sem IV) academic year.	
ann	<ul> <li>6. The project should be divided into the following parts:-</li> <li>a) Certification of completion of Project Work</li> <li>b) Acknowledgement</li> <li>c) Introduction</li> <li>d) Review of Related Literature</li> </ul>	
	e) Aims and Objectives f) Plan of work g) Material and Methods h) Results i) Discussion	
	j) Bibliography 7. The project will be assessed.	

#### GUIDELINE FOR THE ASSESMENT OF PROJECT WORK

- 1. The practical 403 of Sem IV (Course Code No. RUSBCHP403) shall be exclusively devoted for the project.
- 2. Each student will complete the project (2 copies) and get both the copies certified by the guiding teacher and the Head of Dept. (HOD) by January of SYBSc (Sem IV) academic year.
- 3. One copy of the certified project will be submitted to the Department; while the other copy will be retained by the students for his/ her personal record.
- 4. The candidate is required to present the Research Project to the examiner followed by Viva- Voce examination based on the project work by the examiner.
- 5. The following Marking Scheme shall be considered while assessing the project work

	Particular	Marks
a)	Project Work (Contents Submitted in the bound form)	10
b)	Presentation of Project Work to examiner	10
c)	Viva- voce Exam based on Project Work	05
d)	Attendance	05
	TOTAL	30

#### **References:**

#### **Semester III**

- 1) Textbook of Medical Physiology Gyton and Hall, Elsevier publishers
- 2) David L. Nelson, Michal M. Cox, Lehniger Principles of Biochemistry, W. H. Freeman & company, New york, 4th edition
- 3) Murry, R. K. & other, Harper's Biochemistry, Appleton & Lange, california, 21st edition
- 4) J. L. Jain, Fundamentals of Biochemistry, S. Chand & company, 2005 edition
- 5) Dr. A.C. Deb, Fundamentals of Biochemistry, New central book agency (P) Ltd., 8th edition
- 6) U. Satyanarayanan, Biochemistry, Books & allied (P) Ltd., Kolkata, 3rd edition
- 7) Murry, R. K. & other, Harper's Biochemistry, Appleton & Lange, california, 21st edition.
- 8) J.B.Russel . Genetics
- 9) BenaminLewin, Gene VII, Oxford University Press
- 10) M.W. Strickberger, Genetics
- 11) William & Wilson

- 1) Microbiology by Stanier
- 2) Stanier, Microbiology Pelzer, Essentials of Microbiology
- 3) Friedfielder, Microbial genetics
  - A. H. Patel, Industrial microbiology, Macmillan India Ltd.
- 4) L. E. Casida, Industrial microbiology, New age international publishers
- 5) F.S.K Brar, Essentials of Pharmacology, S. Chand Publisher
- 6) Upadhyay, Biophysical chemistry, Himalayan Publisher
- 7) Keith Wilson & John Walker, Practical Biochemistry, principle and technique, Cambridge University, 5th edition
- 8) Biotechnology, by Primrose, Dube
- 9) Biotechnology, Jodgan
- 10) U. Satyanarayanan, Biotechnology, Books & allied (P) Ltd., Kolkata

#### **MODALITY OF ASSESSMENT**

#### **Theory Examination Pattern:**

B) Internal Assessment - 40% :40 marks.

Sr No	Evaluation type	Marks
	<b>~ \</b>	
1	One Assignment/presentation/Quiz	20
2	One class Test (multiple choice questions / subjective)	20

#### B) External examination - 60 %

#### Semester End Theory Assessment - 60 marks

- iii. Duration These examinations shall be of **2 hours** duration.
- iv. Paper Pattern:
  - 3. There shall be **04** questions each of **15** marks. On each unit there will be one question &last question will be based on all 3 units.
  - 4. All questions shall be compulsory with internal choice within the questions.

Questions	Options	Marks	Questions on
Q.1)A)	Any 2 out of 3	04	Unit I
Q.1)B)	Any 2 out of 3 OR Any 4 out of 6	06	
Q.1)C)	Any 1 out of 2	05	

Q.2)A)	Any 2 out of 3	04	Unit II
Q.2)B)	Any 2 out of 3 OR Any 4 out of 6	06	
Q.2)C)	Any 1 out of 2	05	
Q.3)A)	Any 2 out of 3	04	Unit III
Q.3)B)	Any 2 out of 3 OR Any 4 out of 6	06	
Q.3)C)	Any 1 out of 2	05	. (
Q.4)A)	Any 1 out of 2	04	Unit I, II, III
Q.4)B)	Any 1 out of 2	04	
Q.4)C)	Any 1 out of 2	04	
Q.4)D)	Any 3 out of 5	03	,5

#### **Practical Examination Pattern:**

#### (A) Internal Examination:

Heading	Practical I, II & III
Journal	05
Test	. 15
Total	20

#### (B) External (Semester end practical examination):

Particulars	Practical I, II & III
Laboratory work	25
Viva	5
Total	30

#### PRACTICAL BOOK/JOURNAL

The students are required to present a duly certified journal for appearing at the practical examination, failing which they will not be allowed to appear for the examination.

In case of loss of Journal and/ or Report, a Lost Certificate should be obtained from Head/ Co-ordinator / Incharge of the department; failing which the student will not be allowed to appear for the practical examination.

## Overall Examination and Marks DistributionPattern Semester – III & IV

Course	RUSBCH301			RUSBCH302			RUSBCH303			Grand
										Total
	Internal	External	Total	Internal	External	Total	Internal	External	Total	
Theory	40	60	100	40	60	100	40	60	100	300
Practicals	20	30	50	20	30	50	20	30	50	150

Course	RUSBCH401			RUSBCH402			RUSBCH403			Grand
										Total
	Internal	External	Total	Internal	External	Total	Internal	External	Total	
Theory	40	60	100	40	60	100	40	60	100	300
Practicals	20	30	50	20	30	50	20	30	50	150

Resolution No. AC/II(18-19).2.RUS2

### S.P.Mandali's Ramnarain Ruia Autonomous College



#### Syllabus for T.Y.BSc.

Program: B.Sc.

Course: Biochemistry (RUSBCH)

(Credit Based Semester and Grading System with effect from the academic year 2019–2020)

#### Semester V

<b>Course Code</b>	Unit	Topics	Credits	Lectures			
Paper I - Metabolism & Analytical Techniques - I							
	I	Carbohydrate Metabolism		15			
	II	Amino acid and protein metabolism &		15			
RUSBCH501		Biochemistry of senses	3				
ROSDCII301	III	Bioenergetics & Oxidative	3	15			
		Phosphorylation		13			
	IV	Chromatography		15			
Paper II – Environmental Science & Cell Biology							
RUSBCH502	U	Air	3	15			
	II	Water & Water Treatment		15			
	III Soil & Noise & Environmental Monitoring		15				
	IV	Cell – Cell Communication		15			
Paper III –Advanced Genetics & RDT							
7.0.	I	DNA Replication & Repair	3	15			
RUSBCH503	II	Transcription		15			
	III	Translation	3	15			
	IV	Recombinant DNA Technology (RDT)		15			
Paper IV – Immunology & Pathophysiology - I							
RUSBCH504	I	Human immune system	3	15			
KUSBCH304	II	Antigen & Antibody	J	15			

	III Antigen- Antibody interactions &		15
	Complement system		
	IV Tumour immunology & Apoptosis		15
RUSBCHP501	Practicals based on course in theory –		
KOSDCIII 301	RUSBCH501	4	
RUSBCHP502	Practicals based on course in theory –	7	.0
RUSDCIII 302	RUSBCH502		0
RUSBCHP503	Practicals based on course in theory –		070
KUSBCHF 303	RUSBCH503	1	
RUSBCHP504	Practicals based on course in theory –	4	
KUSBCHF304		)	
	Semester VI		
<b>Course Code</b>	Unit Topics	Cre	dits Lectu

<b>Course Code</b>	Unit	Topics	Credits	Lectures				
Paper I - Metabolism & Analytical Techniques - II								
	Ι	Lipid metabolism		15				
	II	Nucleic Acid Metabolism & Integration of		15				
RUSBCH601		Metabolism	3	13				
ROSBCIIOOI	III	Centrifugation & Protein Purification	3	15				
		techniques						
	IV	Electrophoresis		15				
Paper II – Nutrition & Pharmacology								
	IO	Nutrition & Diet Management		15				
7	И	Vitamins and Co-enzymes & their deficiency		15				
RUSBCH602		disorders	3	13				
ROSDC11002	III	Pharmacokinetics and Bioassay	3	15				
	IV	Therapeutic drugs & Drugs acting on the		15				
7.0.		Haemopoietic system		13				
Paper III – Biostatistics & Bioinformatics								
	I	Biostatistics & descriptive statistics		15				
RUSBCH603	II	Probability & Normal distribution and		15				
		Bioinformatics Hypothesis testing of means & ANOVA		13				
	III			15				
	IV	Hypothesis testing of difference between means		15				

		& Chi-square test		
	P	aper IV – Immunology & Pathophysiology - II		
	Ι	Metabolic disorders		15
	II	TCR, Major Histocompatibility complex		15
RUSBCH604		&Transplant immunology	3	13
	III	Virology & AIDS		15
	IV	Ageing, Alzheimer's and Vaccines		15
RUSBCHP601	Practio	cals based on course in theory – RUSBCH601	4 0	
RUSBCHP602	Practio	cals based on course in theory –RUSBCH602	14	
RUSBCHP603	Practio	cals based on course in theory –RUSBCH603		
RUSBCHP604	Practio	cals based on course in theory –RUSBCH604	7	

Course Code:RUSBCH Course Title:Biochemistry Academic year 2019-20

## **Learning Objectives:**

The overall goal of this TYBSc course is to familiarize the students to the fields of physiology, metabolism, instrumentation, environmental science, genetics, immunology, nutritional biochemistry, biostatistics, pharmacology and bioinformatics.

## Learning Outcomes:

Upon completion of the TYBSc course, the students would understand the following:

- 1) Metabolism (Carbohydrates, amino acids & proteins, lipids, nucleic acids) & integration of metabolisms. Bioenergetics & oxidative Phosphorylation which will enable them to understand the energetics of various metabolisms.
- 2) Biochemistry of senses topic which includes the study of sensory system of humans.
- 3) Analytical techniques like chromatography, centrifugation, electrophoresis, protein purification techniques, etc
- 4) Environmental Science which include topics like air pollution; water pollution; soil & noise pollution; and Energy, Industrial Pollutants & Environmental Monitoring.

- 5) Nutritional Biochemistry comprising of topics like Nutrition & Diet Management; Vitamins and Co-enzymes & their deficiency disorders.
- 6) Pharmacokinetics and Bioassay; Therapeutic drugs & Drugs acting on haemopoietic systemwill help them to understand pharmacology in more depth.
- 7) The important genetic processes namely, DNA replication, transcription, translation & Recombinant DNA Technology increasing their knowledge of molecular biology.
- 8) Biostatistics which will help them to interpret results and draw conclusions in the research.
- 9) Bio-informatics and applications of computers in Bio-chemistry.
- 10) Basics of human immune system, detailed study of various cells and organs involved.
- 11) Tumour immunology, apoptosis, virology, AIDS, Ageing, Alzheimers, transplant immunology and vaccines which will further increase their understanding of Human immune system in a better way.
- 12) All the practicals have been rearranged in accordance with the theory of each paper at each semester.

The over-all syllabus at the Under-Graduation level has been designed such that the student is well prepared to appear for competitive examinations held all over.

## **Detailed Syllabus**

		SEMESTER V	
Course	Code	Title	Credits
RUSBC	H501	Metabolism & Analytical Techniques - I	3
Unit I C	arbohyo	lrate metabolism	
1.1	Catabo	lism – Cellular location, sequence of reactions, labelling of carbon atoms, and	15 Lectures
	energet	ics of :	
1.1.1	Glycol	ysis (aerobic and anaerobic); - Mechanisms of regulation of glycolysis -	
	Homeo	stasis and Metabolic Control, Metabolic Flux, Metabolic Control Analysis,	
		-Demand Analysis, Mechanisms of Flux Control	
1.1.2	Oxidati	on of pyruvate,	
1.1.3	Krebs	cycle; Glyoxylate pathway;	
1.1.4	Glycog	enolysis – [schematic – no structures, but with enzymes and coenzymes]	
1.2	Anabol	ism – HMP shunt ( Cellular location, sequence of reactions, multifunctional	
	nature)	; Gluconeogenesis, Glycogenesis – [schematic – no structures, but with	
	enzyme	es and coenzymes]	
10			
Unit II	Amino a	acid and protein metabolism & Biochemistry of Senses	
2.1	Amino	acid and protein metabolism	15 Lectures
2.1.1	Reaction	ons of amino acids – Transamination [GOT/GPT and mechanism of	
	transan	nination]; Decarboxylation [His, Trp, Glu, and mechanism of	
		oxylation], Deamination [oxidative – NAD(P) linked dehydrogenases and D	
	& L - A	Amino acid oxidases, Non oxidative – Asp, Cys, Ser]	
2.1.2	Definit	ion – ketogenic and glucogenic amino acids	

2.1.3	Urea cycle – Cellular location, sequence of reactions, Labeling of N atom, transport	
	of NH <sub>3</sub>	
2.2	Biochemistry of Senses –	
2.2.1	Sensory systems, Senses and receptors	
2.2.2	Classification of receptors based on location – Exteroceptors, Interoceptors &	
2 2 2	Proprioceptors	
2.2.3	Classification of receptors based on type of stimulus detected – Mechanoreceptors,	.0
	Photoreceptors, Thermoreceptors, Nociceptors, Chemoreceptors (Gustatory &	00
224	Olfactory), Osmoreceptors	
2.2.4 2.2.5	Auditory system Vision	
2.2.5	Somatosensory	
2.2.0	Somatosensory	
Ilmit III	Piganougating & Ovidative Phagphanylation	
3.1	Bioenergetics & Oxidative Phosphorylation  Bioenergetics	15 Lectures
3.1.1	Concept of free energy; Respiratory electron transport chain – Carriers   basic	13 Lectures
3.1.1	chemistry, redox potentials, orientation on the membrane, sequence); Complexes I to	
	IV; Q cycle in Complex III; Inhibitors of electron transport – Antimycin A,	
	Amytal, Rotenone, CN, Azide, CO	
3.1.2	Malate-Aspartate shuttle and Glycerol phosphate shuttle	
3.2	Oxidative phosphorylation	
	Chemiosmotic hypothesis, Proton motive force; Structure of ATP synthase,	
	Uncoupler- of ETC and Oxidative phosphoryaltion [DNP]	
Unit IV	Chromatography	
4.1	Chromatography	15 Lectures
4.2	Introduction, Concept of partition coefficient, retention time, retention factor	
	Principle, Technique and Applications of the following kinds of chromatography:	
	Partition chromatography ( Paper, GLC)	
	Adsorption Chromatography (TLC and Column)	
	Ion exchange chromatography	
	Gel filtration	
	Affinity chromatography	
	Advanced Chromatographic techniques – HPLC, HPTLC	
	DD A CITICAL C	
	PRACTICALS	
RUSBC	CHP501   Practicals based on course in theory –RUSBCH501	2
	Identification of biomolecules	
	2) Determination of the optimum pH of β-Amylase	
	3) Determination of Km value of β-Amylase	
•	4) Study the effect of inhibitor on β-Amylase	
	5) Determination of the activity and specific activity of β-Amylase	
7.0	from sweet potato	
	6) Estimation of glucose by Benedict's method	
	7) Separation of sugars by circular paper chromatography	
	8) Separation of plant pigments by Adsorption Column	
	Chromatography	

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Correct	Codo	SEMESTER V	Cradita
Course RUSBO		Title Environmental Science & Cell Biology	Credits 3
Unit I A		Environmental science & Cen Biology	
1.1	Atmos	phere	15 Lectures
1.1.1		osition and structure of atmosphere	
1.1.2		es, ions and radicals in the atmosphere	
1.1.3	Chemi	cal and photochemical reactions in the atmosphere [reactions of oxygen, ozone,	
		r dioxide, nitrogen oxide and organic compounds]	
1.2		llutants – CO, Oxides of Nitrogen, SO <sub>2</sub> , hydrocarbons and photochemical smog,	
		house gases, suspended particulate matter[ sources and effect of], depletion of	
	ozone		
IIni4 II	Weter	P. Water Treatment	
2.1	Water	& Water Treatment	15 Lectures
2.1.1		sphere- characteristics and the water cycle	13 Lectures
2.1.1	•	Pollution	
2.2.1	-	c pollutants[pesticides, insecticides, detergents, oil spills, toxic organic	
2.2.2	chemic	-1	
2.3		nic pollutants [heavy metals – Hg, Pb, As, Cd] Thermal pollution of water	
2.3.1		treatment	
10		a for water purity, Water purification [preliminary, primary, secondary, tertiary-	
	chlorin	ation, ion exchange]	
Unit III	   Soil N	oise & Environmental monitoring	
3.1	Soil	oise & Environmental monitoring	15 Lectures
3.1.1		osition of soil,	15 Lectures
3.1.2	-	en cycle	
3.1.3	_	of soil pollution – acidification, agrochemical pollution, salinization, and	
		nination by metalliferous wastes	

Classifica Causes an Environn Approach Significa Remote S Cell – Ce Introduct Cell Junc Occludin Anchorin attachme Commun	ell Communication ion, Significance of cell communication etions & its classification — g junction (tight junctions, septate junctions) ag junction (Actin filament attachment sites & Intermediate filament nt sites),	15 Lect
Causes at Environn Approach Significa Remote S  Cell – Cel Introduct Cell Junc Occludin Anchorin attachme Commun	nd consequences of Noise pollution nental monitoring nes used to monitor the environment-air, water and soil. [Principles and nce only. Protocols for each factor – not required ] Sensing  Ell Communication ion, Significance of cell communication ctions & its classification – g junction (tight junctions, septate junctions) ag junction (Actin filament attachment sites & Intermediate filament nt sites),	15 Lect
Environn Approach Significa Remote S  Cell – Ce Introduct Cell Junc Occludin Anchorin attachme Commun	mental monitoring mes used to monitor the environment-air, water and soil. [Principles and note only. Protocols for each factor – not required ]  Sensing  Ell Communication  ion, Significance of cell communication  etions & its classification –  g junction (tight junctions, septate junctions)  ag junction (Actin filament attachment sites & Intermediate filament nt sites),	15 Lect
Approach Significa Remote S  Cell – Cell Introduct Cell Junco Occludin Anchorin attachme Commun	nes used to monitor the environment-air, water and soil. [Principles and nce only. Protocols for each factor – not required ]  Sensing  Ell Communication  ion, Significance of cell communication  etions & its classification –  g junction (tight junctions, septate junctions)  ag junction (Actin filament attachment sites & Intermediate filament nt sites),	15 Lect
Cell – Cell Introduct Cell Junc Occludin Anchorin attachme Commun	Sensing  Cell Communication  ion, Significance of cell communication  tions & its classification —  g junction (tight junctions, septate junctions)  ag junction (Actin filament attachment sites & Intermediate filament  nt sites),	15 Lect
Remote S  Cell – Ce Introduct Cell Junc Occludin Anchorin attachme Commun	Sensing  Cell Communication  ion, Significance of cell communication  etions & its classification —  g junction (tight junctions, septate junctions)  ag junction (Actin filament attachment sites & Intermediate filament  nt sites),	15 Lect
Cell – Cell Introduct Cell Junc Occludin Anchorin attachme Commun	ell Communication ion, Significance of cell communication etions & its classification — g junction (tight junctions, septate junctions) ag junction (Actin filament attachment sites & Intermediate filament nt sites),	15 Lect
Introduct Cell Junc Occludin Anchorin attachme Commun	ion, Significance of cell communication etions & its classification — g junction (tight junctions, septate junctions) ag junction (Actin filament attachment sites & Intermediate filament nt sites),	15 Lect
Introduct Cell Junc Occludin Anchorin attachme Commun	ion, Significance of cell communication etions & its classification — g junction (tight junctions, septate junctions) ag junction (Actin filament attachment sites & Intermediate filament nt sites),	15 Lect
Cell Junc Occludin Anchorin attachme Commun	etions & its classification — g junction (tight junctions, septate junctions) ag junction (Actin filament attachment sites & Intermediate filament ant sites),	15 Lect
Occludin Anchorin attachme Commun	g junction (tight junctions, septate junctions) g junction (Actin filament attachment sites & Intermediate filament nt sites),	
Anchorin attachme Commun	ng junction (Actin filament attachment sites & Intermediate filament nt sites),	
attachme Commun	nt sites),	
Commun		
$C_{\alpha}$ 11 $C_{\alpha}$	icating junction (Gap junction, Chemical synapses, Plasmodesmata)	
	ell Adhesion	
Mechanis	sm of Adhesion – Junctional & Non-junctional	
Cadherin	s mediated Ca <sup>2+</sup> -dependent Cell - Cell Adhesion	
	PRACTICALS	
HP502	Practicals based on course in theory –RUSBCH502	2
	7) Determination of the Chemical Oxygen Demand of Water/	
	Effluent by the Potassium Dichromate method	
	8) Determination of potability of water by conducting a coliform	
	count	
	CHP502	Cadherins mediated Ca <sup>2+</sup> -dependent Cell - Cell Adhesion  PRACTICALS

			(60)
		SEMESTER V	
Course		Title	Credits
RUSBO		Advanced Genetics & RDT  lication & Repair	3
1.1.1		tion of DNA (in prokaryotes)	15 Lectures
1.1.2		of DNA replication : Semi-conservative, Dispersive & Conservative	15 Lectures
1.1.3		of DNA replication: Theta & rolling circle	
1.1.4		es (pol I, II and III) and accessory proteins	
1.1.5		ism of semi-conservative replication (Initiation, elongation & termination)	
1.1.6		lutations, mutagens	
1.2		pair: Direct, Photoreactivation, O6 methyl guanine DNA methyl transferase,	
	Excision	n repair, Mismatch repair, Recombination repair, SOS-error prone repair	
	<u>Γranscr</u>		
2.1		iption in prokaryotes	15 Lectures
2.1.1	Prokary	otic RNA polymerase and promoter; Transcription unit, Upstream regulatory	
2.1.2	sequenc	ism of RNA transcription: Initiation, elongation and termination (Type I & II)	
2.1.2		rative overview of transcription in prokaryotes & eukaryotes	
2.3		ing of tRNA, rRNA, mRNA (prokaryotes and eukaryotes)	
2.4		t of split genes, reverse transcription.	
2.4 2.5	, .	Inhibitor-Rifampicin, Actinomycin D	
Unit III			
3.1		ion (protein biosynthesis) in prokaryotes	15 Lectures
3.1.1		code, mechanism of translation: Activation of amino acids, chain initiation,	
	_	on & termination	
3.2	_	ypothesis	
3.3	Post trai	nslational modifications of proteins (proteolytic cleavage, acylation,	

4.2 Tools for RDT (a) Enzymes- Restriction endonucleases, ligases, terminal transferases, reverse transcriptase: (b) Cloning and Expression Vectors- Plasmid, pBR 322, PUC-19, Bacteriophage – Lambda phage; Cosmid; Artificial Chromosomes(BAC and YAC); Shuttle vectors; (c) Probes- DNA probes 4.3 Applications of RDT- Agriculture (Bt Cotton); Medicine (Insulin); GM food 4.4 Isolation of gene: Gene library and c-DNA library; Southern blot; Northern blot; 4.5 Gene Transfer: Transfection, Electroporation, Microinjection, Liposome, Microprojectile (in brief) Selection and screening- Antibiotic and colony hybridization DNA Amplification by PCR  PRACTICALS  RUSBCHP503 Practicals based on course in theory –RUSBCH503  1) Estimation of glucose by the Folin-Wu method 2) Estimation of Calcium by the Erichrome Black T- EDTA method 4) Estimation of Magnesium by the Frichrome Black T- EDTA method 5) Estimation of phosphorus by Fiske Subbarow method 6) Estimation of DNA by the Diphenylamine method 7) Isolation and spooling of DNA from onion/moong 8) Estimation of UV absorption of nucleic acids& proteins	Introduction of RDT   Tools for RDT (a) Enzymes- Restriction endonucleases, ligases, terminal transferases, reverse transcriptase: (b) Cloning and Expression Vectors- Plasmid, pBR 322, PUC-19, Bacteriophage – Lambda phage; Cosmid; Artificial Chromosomes(BAC and YAC); Shuttle vectors; (c) Probes- DNA probes   Applications of RDT- Agriculture (Bt Cotton); Medicine (Insulin); GM food   Isolation of gene: Gene library and c-DNA library; Southern blot; Northern blot; Gene Transfer: Transfection, Electroporation, Microinjection, Liposome, Microprojectile (in brief)   Selection and screening- Antibiotic and colony hybridization   DNA Amplification by PCR   PRACTICALS      RUSBCHP503   Practicals based on course in theory —RUSBCH503     1			rylation, methylation, glycosylation), Protein targeting s of translation	
4.1 Introduction of RDT 4.2 Tools for RDT (a) Enzymes- Restriction endonucleases, ligases, terminal transferases, reverse transcriptase: (b) Cloning and Expression Vectors- Plasmid, pBR 322, PUC-19, Bacteriophage – Lambda phage; Cosmid; Artificial Chromosomes(BAC and YAC); Shuttle vectors; (c) Probes- DNA probes Applications of RDT- Agriculture (Bt Cotton); Medicine (Insulin); GM food Isolation of gene: Gene library and c-DNA library; Southern blot; Northern blot; Gene Transfer: Transfection, Electroporation, Microinjection, Liposome, Microprojectile (in brief) Selection and screening- Antibiotic and colony hybridization DNA Amplification by PCR  PRACTICALS  RUSBCHP503 Practicals based on course in theory –RUSBCH503  1) Estimation of glucose by the Folin-Wu method 2) Estimation of glucose by the GOD-POD method 3) Estimation of Calcium by the Erichrome Black T- EDTA method 4) Estimation of Magnesium by the Erichrome Black T- EDTA method 5) Estimation of DNA by the Diphenylamine method 7) Isolation and spooling of DNA from onion/moong 8) Estimation of UV absorption of nucleic acids& proteins	Introduction of RDT   Tools for RDT (a) Enzymes- Restriction endonucleases, ligases, terminal transferases, reverse transcriptase: (b) Cloning and Expression Vectors- Plasmid, pBR 322, PUC-19, Bacteriophage – Lambda phage; Cosmid; Artificial Chromosomes(BAC and YAC); Shuttle vectors; (c) Probes- DNA probes   Applications of RDT- Agriculture (Bt Cotton); Medicine (Insulin); GM food   Isolation of gene: Gene library and c-DNA library; Southern blot; Northern blot; Gene Transfer: Transfection, Electroporation, Microinjection, Liposome, Microprojectile (in brief)   Selection and screening- Antibiotic and colony hybridization   DNA Amplification by PCR   PRACTICALS      RUSBCHP503   Practicals based on course in theory —RUSBCH503     1				
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1.1.1 Int 1.1.2 Ct 1.1.3 Ac 1.1.4 Hu 1.2 Or 1.2.1 Pr 1.2.2 Se 1.3 Cc 1.3.1 Ly Gr 1.3.2 Cl 1.3.3 Cy ali  Unit II An 2.1 Ar im	nate immunity – Anatomical barriers, physiological barriers, naracteristics of Inflammation, phagocytosis [no mechanism] daptive immunity – Active & Passive amoral& Cell mediated immunity gans of the immune system imary lymphoid organs: Thymus, Bone marrow econdary lymphoid organs: Lymphatic system, Lymph nodes, Spleen, MALT. ells of the immune system: emphocytes – B cells and T cells, Natural killer cells – Mononuclear phagocytes, ranulocytes, Antigen presenting cells. onal selection & immunologic memory. etokines: biological functions of IL1, tumor necrosis factor alpha, interferon – oha, IL2, interferon-gamma.  **Integers: Antibody**  **Integers: Antigenecity, immunogenecity, epitope, factors determining**	
1.1.1 Int 1.1.2 Ct 1.1.3 Ac 1.1.4 Ht 1.2 Or 1.2.1 Pr 1.2.2 Se 1.3 Cc 1.3.1 Ly Gr 1.3.3 Cy ali  Unit II An  2.1 Ar Ar	nate immunity – Anatomical barriers, physiological barriers, naracteristics of Inflammation, phagocytosis [no mechanism] daptive immunity – Active & Passive amoral& Cell mediated immunity rgans of the immune system imary lymphoid organs: Thymus, Bone marrow econdary lymphoid organs: Lymphatic system, Lymph nodes, Spleen, MALT. ells of the immune system: ramphocytes – B cells and T cells, Natural killer cells – Mononuclear phagocytes, ranulocytes, Antigen presenting cells. onal selection & immunologic memory. vtokines: biological functions of IL1, tumor necrosis factor alpha, interferon – pha, IL2, interferon–gamma.  **tigen-Antibody**  *tigen-Antibody**  *tigens: Antigenecity, immunogenecity, epitope, factors determining imunogenecity, Haptens, adjuvants. atibodies: Fine structure of immunoglobulin, Antibody mediated functions, intibody classes, Monoclonal antibodies.	
1.1.1 Ini 1.1.2 Ch 1.1.3 Ac 1.1.4 Hu 1.2 Or 1.2.1 Pr 1.2.2 Se 1.3 Cc 1.3.1 Ly Gr 1.3.2 Cl 1.3.3 Cy ali  Unit II An 2.1 Ar im 2.2 Ar Ar 2.3 Ar	nate immunity – Anatomical barriers, physiological barriers, naracteristics of Inflammation, phagocytosis [no mechanism] daptive immunity – Active & Passive amoral& Cell mediated immunity rgans of the immune system imary lymphoid organs. Thymus, Bone marrow econdary lymphoid organs: Lymphatic system, Lymph nodes, Spleen, MALT. ells of the immune system: //mphocytes – B cells and T cells, Natural killer cells – Mononuclear phagocytes, ranulocytes, Antigen presenting cells. onal selection & immunologic memory. // tokines: biological functions of IL1, tumor necrosis factor alpha, interferon – pha, IL2, interferon-gamma.  **Itigen- Antibody** Intigens: Antigenecity, immunogenecity, epitope, factors determining munogenecity, Haptens, adjuvants. Intibodies: Fine structure of immunoglobulin, Antibody mediated functions, intibody classes, Monoclonal antibodies. Intibody diversity: Multigene organization of immunoglobulin genes – Lambda ,	
1.1.1 Ini 1.1.2 Ch 1.1.3 Ac 1.1.4 Hu 1.2 Or 1.2.1 Pr 1.2.2 Se 1.3 Cc 1.3.1 Ly Gr 1.3.2 Cl 1.3.3 Cy ali  Unit II An 2.1 Ar 2.2 Ar 2.3 Ar ka	nate immunity – Anatomical barriers, physiological barriers, naracteristics of Inflammation, phagocytosis [no mechanism] daptive immunity – Active & Passive amoral& Cell mediated immunity rgans of the immune system imary lymphoid organs. Thymus, Bone marrow acondary lymphoid organs: Lymphatic system, Lymph nodes, Spleen, MALT. The sells of the immune system:  Amphocytes – B cells and T cells, Natural killer cells – Mononuclear phagocytes, ranulocytes, Antigen presenting cells.  Conal selection & immunologic memory.  Atokines: biological functions of IL1, tumor necrosis factor alpha, interferon – pha, IL2, interferon-gamma.  Atigen- Antibody  Intigens: Antigenecity, immunogenecity, epitope, factors determining munogenecity, Haptens, adjuvants.  Intibodies: Fine structure of immunoglobulin, Antibody mediated functions, intibody classes, Monoclonal antibodies.  Intibody diversity: Multigene organization of immunoglobulin genes – Lambda, ppa & heavy chain	
1.1.1 Int 1.1.2 Cr 1.1.3 Ac 1.1.4 Hr 1.2 Or 1.2.1 Pr 1.2.2 Se 1.3 Cc 1.3.1 Ly Gr 1.3.2 Cl 1.3.3 Cy ali  Unit II An 2.1 Ar ka 2.3 Ar ka 2.3.1 Li	nate immunity – Anatomical barriers, physiological barriers, naracteristics of Inflammation, phagocytosis [no mechanism] daptive immunity – Active & Passive amoral& Cell mediated immunity rgans of the immune system imary lymphoid organs. Thymus, Bone marrow econdary lymphoid organs: Lymphatic system, Lymph nodes, Spleen, MALT. ells of the immune system: //mphocytes – B cells and T cells, Natural killer cells – Mononuclear phagocytes, ranulocytes, Antigen presenting cells. onal selection & immunologic memory. // tokines: biological functions of IL1, tumor necrosis factor alpha, interferon – pha, IL2, interferon-gamma.  **Itigen- Antibody** Intigens: Antigenecity, immunogenecity, epitope, factors determining munogenecity, Haptens, adjuvants. Intibodies: Fine structure of immunoglobulin, Antibody mediated functions, intibody classes, Monoclonal antibodies. Intibody diversity: Multigene organization of immunoglobulin genes – Lambda ,	

3.1		tibody interactions & Complement system ibody interactions	15 Lecture
3.1.1	_	ed, antibody affinity, antibody avidity.	13 Lecture
3.1.2	*	reactions – Oudins, Ouchterlony	
3.1.3		reactions: Blood typing, bacterial agglutination,	
3.1.4		tination, agglutination inhibition, Coomb's test.	_ (
3.1.5		rophoresis: Principles of Radioimmunoassay, ELISA,	
		escence, Western Blotting	
3.2	Complement		
3.2.1		of complement;	
3.2.2		activation – Classical, Alternate &Lectin pathway; formation of	
	membrane att		
3.2.3	Biological co	nsequences of complement activation.[in brief]	
Unit I		nunology & Apoptosis	
4.1	Physiology of	Tumourous cells	15 Lecture
4.2	Carcinogens:	Types (Physical, Chemical and Biological); Environmental Factor.	
4.3	Role of p53, o	oncogenes and Tumour suppressor genes	
4.3.1	Conversion of	f proto-oncogenes to oncogenes	
4.4	Cancer therap	y (Chemo – purine, pyrimidine and folate analogs)	
4.5	Apoptosis –		
4.5.1		apoptotic cells	
4.5.2		ses in apoptosis	
4.5.3		Intrinsic & Extrinsic pathway)	
		PRACTICALS	
RUSB	CHP504 Pra	acticals based on course in theory –RUSBCH504	2
		Isolation of starch from potato	
		Estimation of percentage purity of starch hydrolysate by Willstatter's	
	2)	method	
	3)	Extraction of lipid from oil seeds by the Cold Percolation Method	
		Determination of Blood groups	
	3)	Determination of the Haemoglobin content by the	
		Sahli'shaemoglobinometer	
		Making of pH paper using anthocyanin extracted from vegetables	
	[7]	Use of anthocyanin as pH indicator	
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Composition	0/1608
SEMESTER VI	
Course Code Title	Credits
RUSBCH601   Metabolism & Analytical Techniques - II	3
Unit I Lipid metabolism	1.57
1.1 Lipid metabolism	15 Lectures
1.1.1 Catabolism - Knoop's experiment, Beta oxidation of even carbon saturated fatty	
acids (C4 to C20) Energetics of fatty acid oxidation  1.2 Anabolism – Fatty acid biosynthesis (palmitic acid), Ketone body formation,	
utilization, and the physiological significance of Ketone bodies in Diabetes mellitu	18
Starvation, Pregnancy and Alcoholism	10,
1.3 Metabolism of Cholesterol: Catabolism, Anabolism & Transport	
Unit II Nucleic Acid Metabolism & Integration of Metabolism	
2.1 Metabolism of Purine and pyrimidine	15 Lectures
2.1.1 Biosynthesis and degradation	
2.1.2 Salvage pathway	
2.1.3 Inhibitors	
2.2 Integration of metabolism	
2.2.1 Integration of major metabolic pathways of energy metabolism	
2.2.2 Organ specialization and metabolic integration –Liver, Adipose tissues, Skeletal muscle, Brain, Kidney	
2.2.3 Metabolism of starvation - Liver, Adipose tissues, Skeletal muscle, Brain	
1.2.3 Memorism of starvation - Liver, Adipose dissues, Skeletai musele, Biam	
Unit III Centrifugation & Protein Purification techniques	
3.1 Centrifugation	15 Lectures
3.1.1 RCF, RPM and derivation of an equation relating the two; Nomogram;	
Sedimentation coefficient	
3.1.2 Types and applications of centrifuges – Clinical, High speed, Ultra centrifuge -	

	preparative and analytical.	
3.1.3	Types of centrifugation and its applications—Differential, Rate zonal, Isopycnic	
	(Centrifugation with and without density gradients)	
3.2	Protein purification techniques	
3.2.1	Methods of Cell rupture – solid shear, liquid shear, high pressure, ultrasound, osmotic	
	shock, chemical treatment (enzyme, organic solvent), temperature, Cell bomb,	
	homogenizer, Beads method	
3.2.2	Protein isolation (Selection of source, methods, stabilization, assay)	AK
3.2.3	Factors affecting protein solubility – Salt Concentration, organic solvents, pH,	
	crystallization	0,9
3.2.4	Molecular filtration of proteins	
Unit IV	7 Electrophoresis	
4.1	Basic principle of electrophoresis, factors affecting rate of electrophoresis, concept of	15 Lectures
	electro-osmotic flow	
4.2	Support Media used in electrophoresis- Agarose, polyacrylamide, Cellulose acetate	
4.3	Electrophoresis of proteins - Concept of discontinuous buffer system, SDS PAGE,	
	Native PAGE, Gradient gel, Isoelectric focusing of gel, 2D gel	
4.4	Detection of protein in gel- CBB, Silver staining, Zinc staining	
	Electrophoresis of Nucleic acid (DNA & RNA) -AGE, PFGE	
4.5	Detection of Nucleic acid in gel- Ethidium bromide, syber green	

	PRACTICALS	
RUSBCHP601	Practicals based on course in theory RUSBCH601	2
	Separation of amino acids by circular paper chromatography	
	2) Determination of the optimum pH of Urease	
	3) Determination of Km of Urease	
	4) Determination of the optimum pH of pectinesterase	
	5) Determination of Km of pectinesterase	
	6) Estimation of lactose by Cole's ferricyanide method	
	7) Estimation of Glucose Iodometrically	
	8) Separation of plant pigments/Oils by Thin Layer Chromatography	
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Cour	se Code	Title	Credits
	BCH602	Nutrition & Pharmacology	3
		& Diet Management	
1.1 1.1.1 1.1.2 1.2 1.3 1.4.1	Nutrient Mg, Na, Nutritio Energy Anthrop	es of nutrition  as [Proximate principles, vitamins and minerals (macro and micro- role of Ca,  and Fe, Zn)], Dietary fibre.  anal status [malnutrition (protein energy and protein calorie) and over nutrition]  Assessment - RQ, BMR  cometry - BMI, Waist:hip ratio  anagement	15 Lectures
1.4.1	Dietary	Management in :- Obesity, Starvation, Diabetes Mellitus, Hypertension, Peptic Obstructive Jaundice	
		s and Coenzymes & their deficiency disorders	
2.1	role and Chemist	Dluble vitamins (Chemistry of the vitamin & its coenzyme form, Biochemical disorders) –Thiamin, Riboflavin, Niacin, Pyridoxine, Biotin, Lipoic acid:- cry of the Vitamin and its coenzyme form [structure not to be done, only group din its activity]	15 Lectures
2.2.1	Fat solu Biocher Vitamin	d in its activity] ble vitamins A,D,E,K (Chemistry of the vitamin & its coenzyme form, nical role and disorders) – A – Chemistry, Wald's Visual cycle and role of Rhodopsin (with structure), cin, cGMP in vision; Deficiency disorders (Night Blindness, Xerosis	

3.1.1	I Pharmac	odynamics, Physicochemical properties of drugs,	15 Lectur			
3.1.2		orption: through-GIT, pulmonary, renal, placental and blood-brain barrier	15 Lectur			
3.1.3		bility and Bioequivalence				
3.1.4		stribution, Metabolism and Excretion				
3.2	_	s: Preclinical and clinical evaluation, Therapeutic drug monitoring				
<u> </u>	Broussays	7 1 1 100 mm cur una ommour o varaantom, 1 morapounto arag momeormg				
Unit IV	Therape	itic drugs & Drugs acting on Haemopoietic System				
4.1	Therapeu	tic drugs: ( Mechanism of action and adverse effects)	15 Lectur			
4.1.1	Anti infla	mmatory – non steroid anti inflammatory NSAID [Ibuprofen], Salicylates –				
	[Aspirins]					
4.1.2	Cardiovascular drugs- CVS [Ca channel blocker-Amlodipine, and Beta blocker –					
	Propreno	lol]				
4.1.3		c – Penicillin and Sulphonamide				
4.1.4		Proton pump blocker –Omeprazole				
4.2		ing on Haemopoietic System				
4.2.1		sm of iron				
4.2.2		py: Oral Iron preparations, Parental Iron preparations, Toxicity of Iron:				
		xamineMesylate				
4.2.3		d (Pteroylglutamic acid): Mode of Action, Therapeutic Uses				
4.2.4		312 (Cyanocobalamin): Mode of Action, Therapeutic Uses				
4.2.5		cobalamin				
4.2.6	Erythropo					
4.2.7		timulating Factors: Filigrastim, Lenograstim, Molgramostim				
4.2.8	_	gulants – Mechanism of Haemostasis				
	Intravenous anticoagulants – Heparin					
	Oral anti	coagulants – Coumarin derivatives & Indanedione derivatives				
		PRACTICALS				
RUSB	CHP602	Practicals based on course in theory –RUSBCH602	2			
		1) Estimation of Fluoride in water by the Alizarin Red method				
		2) Determination of the Dissolved Oxygen content of water/				
		Effluent by the Winkler's Iodometric method				
		3) Determination of the Biological Oxygen Demand of water/				
		Effluent				
	•	4) Determination of the Alkalinity of water/ Effluent				
	7	5) Determination of the Acidity of water/ Effluent				
		6) Estimation of CaCO <sub>3</sub> of soil by Bromothymol Blue Method				
		7) Demonstration of Sohxlet method				
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Course Code       Title       Credits         RUSBCH603       Biostatistics & Bioinformatics       3         Unit I Biostatistics & Descriptive Statistics       1.1 Introduction: scope and applications of biostatistics         1.2       Common statistical terms: Sources, nature and presentation of data; Measurement and scales of measurement       1.3 Descriptive statistics: Measures of central tendency- Mean, Median and mode         1.4       Measures of dispersion- Range, percentiles, variance, SD, Mean deviation,         Unit II Probability, Normal Distribution & Bioinformatics         2.1       Probability       15 Lectures         2.1.1       Concept of probability: definition       15 Lectures         2.1.2       Normal distribution and normal curve, Asymmetric distribution       15 Lectures         2.1.3       Statistical problems based on the above concepts       Bioinformatics:						
RUSBCH603   Biostatistics & Bioinformatics   3	Cours	SEMESTER VI	Credits			
Unit I Biostatistics & Descriptive Statistics  1.1 Introduction: scope and applications of biostatistics 1.2 Common statistical terms: Sources, nature and presentation of data; Measurement and scales of measurement 1.3 Descriptive statistics: Measures of central tendency- Mean, Median and mode 1.4 Measures of dispersion- Range, percentiles, variance, SD, Mean deviation,  Unit II Probability, Normal Distribution & Bioinformatics  2.1 Probability 2.1.1 Concept of probability: definition 2.1.2 Normal distribution and normal curve, Asymmetric distribution 2.1.3 Statistical problems based on the above concepts 2.2 Bioinformatics:						
1.1 Introduction: scope and applications of biostatistics 1.2 Common statistical terms: Sources, nature and presentation of data; Measurement and scales of measurement 1.3 Descriptive statistics: Measures of central tendency- Mean, Median and mode 1.4 Measures of dispersion- Range, percentiles, variance, SD, Mean deviation,  Unit II Probability, Normal Distribution & Bioinformatics  2.1 Probability 2.1.1 Concept of probability: definition 2.1.2 Normal distribution and normal curve, Asymmetric distribution 2.1.3 Statistical problems based on the above concepts 2.2 Bioinformatics:						
1.2 Common statistical terms: Sources, nature and presentation of data; Measurement and scales of measurement 1.3 Descriptive statistics: Measures of central tendency- Mean, Median and mode 1.4 Measures of dispersion- Range, percentiles, variance, SD, Mean deviation,  Unit II Probability, Normal Distribution & Bioinformatics  2.1 Probability Concept of probability: definition 2.1.2 Normal distribution and normal curve, Asymmetric distribution 2.1.3 Statistical problems based on the above concepts 2.2 Bioinformatics:			15 Lectures			
and scales of measurement  Descriptive statistics: Measures of central tendency- Mean, Median and mode  Measures of dispersion- Range, percentiles, variance, SD, Mean deviation,  Unit II Probability, Normal Distribution & Bioinformatics  2.1 Probability  Concept of probability: definition  Normal distribution and normal curve, Asymmetric distribution  Statistical problems based on the above concepts  Bioinformatics:		Common statistical terms: Sources, nature and presentation of data;				
1.4 Measures of dispersion- Range, percentiles, variance, SD, Mean deviation,  Unit II Probability, Normal Distribution & Bioinformatics  2.1 Probability  Concept of probability: definition  Normal distribution and normal curve, Asymmetric distribution  Statistical problems based on the above concepts  2.2 Bioinformatics:		and scales of measurement				
Unit II Probability, Normal Distribution & Bioinformatics  2.1 Probability 2.1.1 Concept of probability: definition 2.1.2 Normal distribution and normal curve, Asymmetric distribution 2.1.3 Statistical problems based on the above concepts 2.2 Bioinformatics:	1.3					
2.1 Probability 2.1.1 Concept of probability: definition 2.1.2 Normal distribution and normal curve, Asymmetric distribution 2.1.3 Statistical problems based on the above concepts 2.2 Bioinformatics:	1.4	Measures of dispersion- Range, percentiles, variance, SD, Mean dev	iation,			
2.1 Probability 2.1.1 Concept of probability: definition 2.1.2 Normal distribution and normal curve, Asymmetric distribution 2.1.3 Statistical problems based on the above concepts 2.2 Bioinformatics:						
2.1.1 Concept of probability: definition 2.1.2 Normal distribution and normal curve, Asymmetric distribution 2.1.3 Statistical problems based on the above concepts 2.2 Bioinformatics:			[ ·			
<ul> <li>2.1.2 Normal distribution and normal curve, Asymmetric distribution</li> <li>2.1.3 Statistical problems based on the above concepts</li> <li>2.2 Bioinformatics:</li> </ul>			15 Lectures			
<ul> <li>2.1.3 Statistical problems based on the above concepts</li> <li>2.2 Bioinformatics:</li> </ul>						
2.2 Bioinformatics:						
2.2.1 Introduction	2.2.1					
2.2.2 Applications of Bioinformatics in – Sequence analysis, Molecular modeling and			odeling and			
drug designing, Phylogeny/evolution, Ecology & population studies, Medical		drug designing, Phylogeny/evolution, Ecology & population studies	, Medical			
informatics and agriculture.	10					
2.2.3 Introduction to Genomics and Proteomics Databases- Definition & types – Public	2.2.3		- 1			
domain database, Sequence database, Structural database, Motif database, Genome			abase, Genome			
database, Proteome database, Annotated sequence database.		I database Proteome database Annotated sequence database				
	2.2.4		OT DDD CDD			
L-ALIGN, CLUSTAL- X & W, RASMOL,	2.2.4	Full form & function in brief of - GenBank, EMBL, PIR, SWISS PF				
2.2.6 Micro-array analysis-concept and applications	2.2.4 2.2.5	Full form & function in brief of - GenBank, EMBL, PIR, SWISS PF Sequence analysis Tools - Explain the following terms in brief - BL				

Unit II	Hypothesis Testing of Means & ANOVA	•
3.1	Introduction – Hypothesis, Type I and Type II errors, One-tailed and two tailed	15 Lectures
	tests,	
3.2	Hypothesis testing of mean - Z-test, t-test	
3.3	Standard error	
3.4	ANOVA	
3.5	Statistical problems based on the above concepts	
Unit IV	Hypothesis Testing of Difference Between Means & Chi-Square Test	
4.1	Hypothesis testing of difference between population means - Z-test, t-test (Paired	15 Lectures
	and unpaired)	
4.2	Chi-square (Test of population variance, Test of goodness of fit, Test of association),	
	2 x 2 Table, Yates' correction	
4.3	Statistical problems based on the above concepts	
	15	

4.3	Statistica	Il problems based on the above concepts	
		*O,	
		PRACTICALS	
RUSB	CHP603	Practicals based on course in theory –RUSBCH603	
		1) Estimation of proteins by the Folin-Lowry method	
		2) Estimation of Iron by Wong's method	
		3) Study of Monograph of Acetyl Salicylate	
		4) Study of Monograph of Sucrose	
		<ul><li>5) Biostatistics – Problems</li><li>6) Isolation of RNA from Yeast/ Liver</li></ul>	
		<ul><li>6) Isolation of RNA from Yeast/ Liver</li><li>7) Estimation of RNA by the Orcinol method</li></ul>	
		7) Estillation of Rever by the Oremor method	
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	SEMESTER VI	
Cour	se Code Title	Credits
	3CH604 Immunology & Pathophysiology - II	3
	Metabolic disorders	
1.1 1.1.1 1.2 1.3 1.4 1.5	Metabolic disorder Inborn error: With respect to Etiology and Clinical manifestations Carbohydrate Metabolism: Glycogen storage diseases, Wernicke-Korsakoff syndrome, Classical galactosemia, essential fructosuria, hereditary fructose intolerance, Lactose intolerance Protein Metabolism: Albinism, Metabolic disorders of urea cycle Lipid Metabolism: TaySach's disease, Niemann-Pick disease, Wolman disease Nucleic acid Metabolism: Purine metabolism disorders (Gout and its types, Lesch- Nyhan syndrome), Pyrimidine metabolism disorders (Orotic aciduria, Reye's syndrome)	15 Lectures
¥7 • · ==		
	TCR MHC &Transplant Immunology	15.1
2.1 2.1.1 2.1.2 2.1.3	T-cell Receptor, Structure & function of $\alpha\beta$ and $\gamma\delta$ Organization & rearrangement of TCR genes TCR complex (TCR-CD <sub>3</sub> ) T-cell accessory membrane molecules	15 Lectures
2.2 2.2.1 2.3	Major histocompatibility complex MHC polymorphism & organization of MHC genes- class I & class II; Cellular distribution & structure of class I & II molecules; Self MHC restriction of T cells. Role of antigen presenting cells.	
2.4 2.4.1	Transplant immunology: Types of transplant; immunological basis of allograft rejection.	

I Init II	I Virology & AIDS	
3.1	Introduction to virology & General Structure of Virus	15 Lecture
3.1.1	Immune responses against viral infections and immune evasion mechanisms by	13 Lecture
3.1.1	viruses	
3.2	Structure and mechanism of replication in:	
3.2	1. Vaccinia	100
	2. Polio	
	3. Influenza	
	4. Hepatitis	7
3.3	AIDS:	
3.3.1	Structure and genetics basis of AIDS virus.	
3.3.2	Replication of AIDS Virus.	
3.3.3	Diagnosis, Symptoms and Causes of AIDS.	
3.3.4	AIDS Therapy.	
Unit I	V Ageing, Alzheimer's & Vaccines	
4.1	Ageing: Definition of ageing. Molecular changes during ageing. Theories of Ageing.	15 Lecture
4.2	Alzheimer's disease – Disease mechanism, characteristics, cause, diagnosis and	
	management	
4.3	Vaccines	
4.3.1	Active & Passive immunization,	
4.3.2	Types of vaccine –	
	1. Live, attenuated vaccine;	
	2. Subunit vaccine – Toxoids (Antivenins), Triple antigen (DTP), Bacterial	
	polysaccharide capsules, outer membrane proteins, viral glycoproteins, pathogen	
	proteins, synthetic peptides (multivalent subunit vaccines) as vaccines 3. Conjugate vaccine; Anti-idiotype vaccines;	
	4. DNA vaccines; Recombinant vector vaccines	
	4. DIVA vaccines, Recombinant vector vaccines	
	PRACTICALS	
DIICD		2
KUSB	CHP604 Practicals based on course in theory –RUSBCH604	2
	1) Isolation of Casein from milk	
	2) Diagnostic test for Typhoid-Widal Qualitative	
	3) Diagnostic test for Typhoid-Widal Quantitative	
	4) Immunoprecipitation reaction of Antigen and antibody	
	5) Separation of serum proteins by PAGE	
	<ul><li>6) Detection of Proteins by Silver Staining Method</li><li>7) Agarose gel electrophoresis- Chromosomal DNA and Plasmid DNA</li></ul>	
	The Agarosa gal alactrophoragic Chromogomal DNA and Plasmid DNA	I

## References:

- official states of the states 1) Lehninger's- Principles of Biochemistry by David L. Nelson
- 2) Harper's Illustrated Biochemistry by Robert K. Murray
- 3) Biochemistry by Donald Voet
- 4) Biochemistry by Jeremy M. Berg
- 5) Biochemistry (2 Volume Set): The Chemical Reactions of Living Cells by David E. Metzler
- 6) Modern Experimental Biochemistry by Rodney F. Boyer
- 7) Basic Concepts in Biochemistry: A Student's Survival Guide by Hiram F. Gilbert
- 8) Analytical Biochemistry by David Holme
- 9) International Biochem by StryerTymoczko Berg
- 10) Biophysical Chemistry Upadhyay
- 11) Biochemistry by Dr. A.C. Deb
- 12) Essentials of Pharmacotherapeutics by FSK Brara
- 13) Textbook of Medical Biochemistry by M.N. Chatterjea&Ranashinde
- 14) Immunology by Goldsby and Kuby
- 15) iGenetics by Russel
- 16) Gene Biotechnology by Jogdand
- 17) Biostatistics by Arora
- 18) Methods of biostatistics for medical students and research workers by Mahajan, B.K.; Jaypeebrothers publishers.
- 19) General Principles of Biochemical Investigation by William & Wilson
- 20) Environmental Chemistry by A.K.De
- 21) Biotechnology by U.Satyanarayana
- 22) Advance in Biotechnology by Jogdand
- 23) Biochemical Calculation by Segel

- 24) Biochemical Methods by Sadashivam
- 25) Introductory Practical Biochemistry by Sawhney
- 26) Practical Biochemistry by David Plummer
- 27) Bioinformatics- Concepts, Skill and applications by Rastogi, S.C.; Mendiratta, Namita and Rastogi, Parag; C.B.S. Publishers & Distributors
- 28) Genes VIII by Lewin, Benjamin; Pearson Prentice and Hall publishers
- 29) Human nutrition and dietetics by Davidson, S. et al.; Churchill Livingstone Publishers.
- 30) Nutrition and dietetics by Joshi, Shubhangini A.; Tata McGraw and Hill publishers
- 31) Nutrition Science by Srilakshmi, B.; New Age International publishers
- 32) Introductory practical biochemistry by Sawhney, S.K. and Singh, Randhir; Narosa Publishing House
- 33) Biochemical calculation by Segel, Irwin H.; John Wiley & Sons publishers
- 34) Text book of Medical physiology by Guyton, Arthur C. and Hall, John E.; Harcourt Brace & Company Asia Pvt Ltd

## **MODALITY OF ASSESSMENT**

## **Theory Examination Pattern:**

C) Internal Assessment - 40% :40 marks.

Sr No	Evaluation type	Marks
1	One test/assignment/quiz/presentation	20
2	One class Test (multiple choice questions / subjective)	20

#### B) External examination - 60 %

### **Semester End Theory Assessment - 60 marks**

- v. Duration These examinations shall be of **2 hours** duration.
- vi. Paper Pattern:
  - 5. There shall be **05** questions each of **12** marks. On each unit there will be one question & first question will be based on all the 4 units.
  - 6. All questions shall be compulsory with internal choice within the questions.

Questions	Options	Marks	Questions on						
Q.1)	Any 6 out of 8	12	Unit I, II, III, IV						
Q.2)A) Any 2 out of 3 06 Unit I									
Q.2)B)	Any 1 out of 2	06							
Q.3)A)	Any 2 out of 3	06	Unit II						
Q.3)B)	Any 1 out of 2	06							
Q.4)A)	Any 2 out of 3	06	Unit III						
Q.4)B)	Any 1 out of 2	06							
Q.5)A)	Any 2 out of 3	06	Unit IV						
Q.5)B)	Any 1 out of 2	06	15						
ractical Examination F	Pattern:	<b>Y</b>							
(A)Internal Ex	amination								

# Practical Examination Pattern:

# (A)Internal Examination

Heading	Practical I
Journal	05
Test	15
Total	20

# (B) External (Semester end practical examination):

Particulars	Practical 1
Laboratory work	25
Viva	5
Total	30

### PRACTICAL BOOK/JOURNAL

The students are required to present a duly certified journal for appearing at the practical examination, failing which they will not be allowed to appear for the examination.

In case of loss of Journal and/ or Report, a Lost Certificate should be obtained from Head/ Coordinator / Incharge of the department; failing which the student will not be allowed to appear for the practical examination.

# Overall Examination and Marks Distribution Pattern

## Semester - V

Course	RUSB	CH501		RUSB	CH502		Grand		
	67,								Total
	Internal	External	Total	Internal	External	Total			
Theory	40	60	100	40	60	100	200		
Practicals	20	30	50	20	30	50	100		

Course	RUSB	RUSBCH503		RUSB	CH504		Grand Total	
)	Internal	External	Total	Internal	External	Total		
Theory	40	60	100	40	60	100	200	
Practicals	20	30	50	20	30	50	100	

#### Semester - VI

Course	RUSBCH601			RUSBCH602			Grand	
							Total	
	Internal	External	Total	Internal	External	Total		
Theory	40	60	100	40	60	100	200	
Practicals	20	30	50	20	30	50	100	

Course	RUSBCH603		RUSBCH604			Grand		
				. (		Total		
							Total	
	Internal	External	Total	Internal	External	Total		
Theory	40	60	100	40	60	100	200	
Practicals	20	30	50	20	30	50	100	
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